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#### A review of structural modification and biological activities of oleanolic acid

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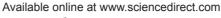
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•Review•

# A review of structural modification and biological activities of oleanolic acid

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[ABSTRACT] Oleanolic acid (OA), a pentacyclic triterpenoid, exhibits a broad spectrum of biological activities, including antitumor, antiviral, antibacterial, anti-inflammatory, hepatoprotective, hypoglycemic, and hypolipidemic effects. Since its initial isolation and identification, numerous studies have reported on the structural modifications and pharmacological activities of OA and its derivatives. Despite this, there has been a dearth of comprehensive reviews in the past two decades, leading to challenges in subsequent research on OA. Based on the main biological activities of OA, this paper comprehensively summarized the modification strategies and structure-activity relationships (SARs) of OA and its derivatives to provide valuable reference for future investigations into OA.

[KEY WORDS] Oleanolic acid; Derivatives; Pentacyclic triterpenoid; Structural modification; Biological activities.

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#### Introduction

Oleanolic acid (OA), a pentacyclic triterpenoid, exists widely in the plant kingdom, either as a free molecule or in a glycosidic form. Its chemical structure is depicted in Fig. 1. OA is abundant in plants, such as leaf gallbladder and Ligustrum lucidum. It is an isomer of ursolic acid (UA), with the sole distinction being the shift of a methyl group from the C-20 to the C-19 position in OA. Both OA and UA exhibit similar biological activities. OA is known for its extensive pharmacological effects, such as antitumor [1], anti-HIV [2], antibacterial [3], anti-inflammatory [4], hepatoprotective [5], antidiabetic [6], anti-atherosclerotic properties [7]. However, literature frequently cites OA's limited activity, with an absolute oral bioavailability of just 0.7%, attributed to its low permeability and water solubility [8]. Malabsorption and extensive metabolic clearance may be responsible for the low oral bioavailability of OA [9]. Structural modifications of OA primarily target the C-3 hydroxyl group, A and C rings, the C-12, 13 double

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bond, and the C-28 carboxyl group. The main purpose of these modifications is to increase its activities, improve its bioavailability, reduce the side effects of itself or its derivatives, and improve its selectivity or targeting. Design strategies include combination principles, prodrug approaches, and the incorporation of specific pharmacophores. This paper reviews recent advancements in the structural modification of OA and its derivatives, categorized by their biological activities. It summarized the fundamental structure-activity relationship (SAR), which provides a reference for future drug development.

# Structural Modification and Biological Activities of OA

#### Antitumor activity of OA and its derivatives

In 2017, Salvador *et al.* reviewed the latest research on semisynthetic derivatives of oleanane, arbutane, and quinone methyl ether with anticancer activities <sup>[10]</sup>. This review highlighted the progress in understanding the anticancer properties of OA and its derivatives from 2012 to early 2017. The current paper builds upon this foundation, providing an updated overview of the antitumor research related to OA in recent years.

In 2020, WANG et al. took OA as the lead compound, implementing structural modifications such as the transform-



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Fig. 1 Structure of OA.

ation of A ring C-2, C-3 into a ketoxime group and the esterification and amidation of the C-28 carboxyl group. This led to the design and synthesis of OA derivatives 1a-1i [11], as shown in Fig. 2. In vitro pharmacological activity tests revealed that these target compounds exerted a certain inhibitory effect on SGC7901 and A549 cells at specific concentrations. Notably, modifications at the C-28 carboxyl group, either forming esters or amides, significantly enhanced their antitumor activity [11]. It was observed that the formation of an ester at the C-28 carboxyl group generally resulted in increased activity proportional to the length of the ester chain. Conversely, amide formation at this site typically yielded stronger activity than the corresponding ester. Among these compounds, 1j demonstrated the most potent antitumor activity against SGC7901 and A549 cells, with IC50 values of 10.76 and 13.28  $\mu$ mol·L<sup>-1</sup>, respectively [11].

1a: R = 
$$\frac{1}{1}$$
 CH3  $\frac{1}{1}$  C<sub>50</sub> > 50 μmol·L<sup>-1</sup>  $\frac{1}{1}$  CH3  $\frac$ 

Fig. 2 Structures of OA derivatives 1a-1j.

The potential of polar substituents at the C-28 position of OA to enhance its anticancer efficacy has been a topic of interest. In this context, WEI et al. in our research group designed and synthesized a series of novel OA-coupled 1,2,3triazole derivatives by employing a Cu(I)-catalyzed Huisgen 1,3-dipolar cycloaddition reaction, as illustrated in Fig. 3 [12]. The anticancer activities of these compounds against a group of five human cancer cell lines were assessed by the MTT method. Compared with OA, most derivatives exhibited strong inhibitory activities against all five cancer cell types, with some even surpassing 5-fluorouracil (5-FU). For instance, compound 2f possessed a strong inhibitory activity against HT1080 cells with an IC<sub>50</sub> value of 3.51  $\mu$ mol·L<sup>-1</sup>, about sevenfold more effective than 5-FU. Moreover, 2f showed minimal cytotoxicity to most normal cells, indicating its selective inhibition of cancer cell growth. Pharmacological experiments further revealed that 2f significantly induced the apoptosis of HT1080 cells, positioning it as a promising lead candidate for future research. The analysis of SAR indicated that compounds with para-substitution on the aromatic ring were more active than those without substitution or with ortho- or meta-substitution. Additionally, compounds featuring electron-absorbing groups on the aromatic ring generally exhibited higher activity than those with no substitution or electron-donating groups at the same position.

OA derivatives modified by N-benzyl tetrazolium are

known for their potent biological activities, such as inhibition of tumor cell proliferation, cell cycle arrest, induction of cell apoptosis and differentiation, and enhanced hydrogen bond formation with purine, pyrimidine, and other base structural fragments [13]. SUN et al. synthesized compounds 3a-3o, as shown in Fig. 4. With cisplatin as a positive control, this study targeted human gastric cancer MKN-45 cells, human breast cancer MCF-7 cells, and rat glioma C6 cells to assess the in vitro antitumor activity using the MTT method [14]. Particularly, derivative 31, which features a nitro group (-NO<sub>2</sub>) in the ortho-position of the benzene ring, exhibited the most potent antitumor effect. Against MKN-45 and C6 cells, 31 achieved inhibition rates of 50.3% and 69.3%, respectively, which are approximately four and nine times greater than those of OA. Furthermore, when the meta-position of the benzene ring was substituted by a fluorine (-F)

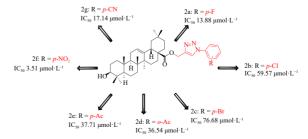


Fig. 3 Structures of OA derivatives 2a-2g.

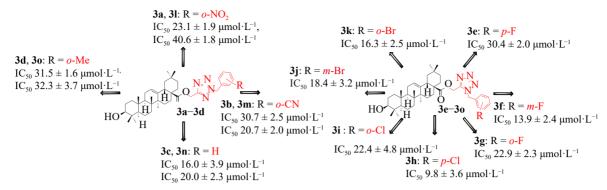


Fig. 4 Structures of OA derivatives 3a-3o.

atom, compound **3f** showed an inhibitory effect on C6 cells approximately ten times that of OA. These results suggest that the conjugation of various benzyl-substituted tetrazoles at the C-28 position of OA significantly enhances its antitumor activity *in vitro*.

Cancer stem cells (CSCs) play an important role in tumorigenesis, tumor recurrence, invasion, and metastasis, and drug resistance [15]. Research has shown that various natural products and their analogs, identified as potential anti-CSC lead compounds [16-19], share a common feature: the presence of Michael acceptors. These active Michael receptors target specific proteins critical in maintaining CSCs. Building on this understanding, researchers synthesized a range of OA derivatives with Michael receptors. The structures of some of these derivatives are illustrated in Fig. 5. Through preliminary in vitro screenings using MTT assays and in vivo toxicity evaluations, OA derivatives 4c and 4d were selected as lead compounds. Both compounds significantly inhibited the proliferation of cancer cells in vivo without exhibiting notable toxic effects on normal tissues. They displayed considerable cytotoxicity against cancer cells, especially melanoma cells (A375 and B16F10). Compound 4d exhibited a more potent inhibitory effect on B16F10 cells, with an IC<sub>50</sub> of  $0.6744 \pm 0.064 \,\mu\text{mol} \cdot \text{L}^{-1}$ . Compound 4c demonstrated superior antiproliferative activity against A375 cells, with an IC<sub>50</sub> of  $0.8274 \pm 0.070 \, \mu \text{mol} \cdot \text{L}^{-1}$ . Importantly, compounds 4c and 4d induced the production of reactive oxygen species (ROS), thereby ablating a variety of CSCs at low concentrations [20].

Triterpenoid saponins containing *N*-acetyl glucosamine, although rare in nature, have demonstrated significant cytotoxicity or antiproliferative activity <sup>[21-26]</sup>. WANG *et al.*, in our research group, previously designed and synthesized a

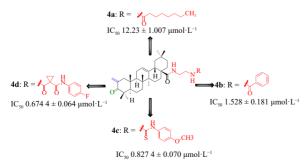


Fig. 5 Structures of OA derivatives 4a-4d.

series of diosgenyl glucosaminides and found that saponins with a cinnamyl group exhibited higher antiproliferative activity on cell lines than other derivatives [27]. Inspired by this result, REN et al. designed and synthesized N-substituted- $\beta$ -D-glucosamine derivatives by a step-by-step glycosylation strategy (Fig. 6). Their cytotoxicity against six different tumor cell lines (HaLa, HepG2, HCT116, MCF-7, A54, and A375-S2) were then evaluated in vitro [28]. The results revealed that most compounds effectively inhibited the growth of at least one tumor cell line at micromolar concentrations. Remarkably, their inhibitory activities against HeLa and HepG2 cells were more pronounced against other tested cells. Among these compounds, 5e stood out for its excellent antiproliferative activity (IC<sub>50</sub> 6.4  $\pm$  0.3 and 7.7  $\pm$  0.2  $\mu$ mol·L<sup>-1</sup> in HeLa and HepG2 cells, respectively), which significantly surpassed the efficacy of 5-FU (IC<sub>50</sub> 63.4  $\pm$  0.8 and 51.5  $\pm$  0.9 μmol·L<sup>-1</sup> in HeLa and HepG2 cells, respectively). At the same time, 5e also showed good selectivity to other cell lines tested.

The poor water solubility of OA leads to low oral absolute bioavailability, and modifications to its hydrophobic scaffold alone cannot significantly improve its solubility <sup>[29]</sup>. Interest has grown in OA derivatives containing *N*-acetylglucosamine (such as compound **6a**, Fig. 7) moiety because of their notable cytotoxicity <sup>[30-33]</sup>. To investigate the effect of glucose isomers on their cytotoxic activity, LIN *et al.* synthesized a series of *N*-acyl, *N*-alkoxycarbonyl, and *N*-alkylcarbamoyl derivatives of 2'-deoxy-glucosyl bearing oleanolic

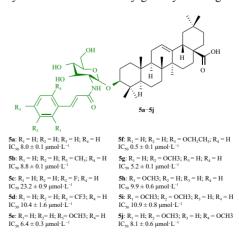


Fig. 6 Structures of OA derivatives **5a-5j**.

RO 
$$\stackrel{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}}{\overset{\dot{H}}}{\overset{\dot{H$$

$$\mathbf{6c}: \mathbf{R} = \begin{array}{c} HO \\ OH \\ HN \\ H_3C \\ \end{array} \qquad \mathbf{6d}: \mathbf{R} = \begin{array}{c} HO \\ OH \\ HN \\ \end{array}$$

 $IC_{s0} \ 11.6 \pm 0.76 \ \mu mol \cdot L^{-1} \quad \ IC_{s0} \ 2.69 \pm 0.03 \ \mu mol \cdot L^{-1}$ 

Fig. 7 Structures of OA derivatives 6a-6d.

saponins and evaluated their activies against HL-60, PC-3, and HT29 tumor cancer cells [34]. They also experimented with varying the length of the carbon chains in these derivatives to improve interaction with target organelles, followed by evaluation of antitumor activity against HL-60, PC-3, and HT29 tumor cells. The SAR analysis revealed that the activity of the 2'-amino group, amides, and urea derivatives followed the order: amino > amide > urea. Some amides and carbamates exhibited stronger growth inhibition than compound 6a, exceeding its cytotoxicity for the first time. The cytotoxic effects of amides, carbamates, and urea compounds with the same carbon chain length on HL-60 cells were almost the same. Increasing the length of the carbon chain enhanced the cytotoxicity of the derivatives, peaking at 11 carbons for amides and nine carbons for carbamates. Compound **6b** emerged as the most cytotoxic against HL-60 cells (IC<sub>50</sub>  $0.76 \pm 0.01 \, \mu \text{mol} \cdot \text{L}^{-1}$ ). Additionally, compounds 6c and 6d, containing galactosamine, displayed almost equivalent cytotoxicity to glucosamine compounds with the same carbon length, indicating that the configuration of C-4 in the sugar chain had no significant effect on cytotoxicity.

To obtain OA derivatives with stronger biological activity and fewer side effects, Medina-O'Donnell *et al.* synthesized a group of OA derivatives [35]. The cytotoxic effects of 18 triterpene derivatives were assayed on three cancer cell lines (B16-F10, HT29, and HepG2) and compared with their impacts on three non-tumor cell lines from the same or similar tissues (HPF, IEC-18, and WRL68). The findings revealed that almost all triterpenic acid diamine conjugates exhibited lower toxicity to non-neoplastic HPF cells, with survival rates ranging from 81% to 94%. The most significant cytotoxic results were observed with diamine conjugates of OA possessing the shortest diamine chain (IC<sub>50</sub> 0.76  $\pm$  0.03 µmol·L<sup>-1</sup>) on the B16F10 cell line, which was around 140-fold more effective than their corresponding precursors (Fig. 8).

Low to intermediate intracellular concentrations of nitric oxide (NO) stimulate oncogenic pathways, while high levels of NO may exert antioncogenic effects. Consequently, numerous studies have investigated OA derivatives with an NO donor for their anticancer properties [36]. The inhibitory effect of OA on the A549 cancer cell line (IC<sub>50</sub> 6.4 μmol·L<sup>-1</sup>) was stronger than that on the other four cancer cell lines (MDA-MB-231, KB, KB-VIN, and MCF-7), After adding a NOdonating nitroxyl group to OA at the end of the ester side chain containing two, four, or six carbons, OA derivatives 8b-8d (Fig. 9) were synthesized, and 8b proved effective against all five cancer cell lines [37]. 8b exhibited stronger cytotoxicity than OA (IC<sub>50</sub> from 4.7 to 6.4 µmol·L<sup>-1</sup>), suggesting that NO donors can increase the potency of OA and broaden its activity spectrum. Compound 8c was less potent, while compound 8d was noncytotoxic (IC<sub>50</sub> > 10  $\mu$ mol·L<sup>-1</sup>) against all five cancer cell lines, suggesting that the linker length may affect cytotoxicity. Compared with OA, 3-acetyl-OA (8a) was somewhat more potent against the five cancer cell lines (A549, MDA-MB-231, KB, KB-VIN, and MCF-7), particularly KB-VIN. Compound 8f showed high and comparable potency against all five cancer cell lines (IC50 from 4.8 to 5.4  $\mu$ mol·L<sup>-1</sup>), while compounds **8e** and **8g** also exhibited cytotoxicity (except for 8e against the KB-VIN cell line, an MDR cell line overexpressing P-glycoprotein), and compound 8f exhibited similar cytotoxicity toward KB and KB-VIN cells. Compounds 8h and 8i were insoluble in DMSO at 10 μmol·L<sup>-1</sup>; thus, their cytotoxicity could not be determined. Compounds 8i, 8k and 8l were not cytotoxic ( $IC_{50} > 10$ μmol·L<sup>-1</sup>). These data indicate that OAc-3 may be a key functional group for enhancing the cytotoxicity of OA-NO hybrids. Compounds 8c and 8f, both containing a nitric oxide donor, markedly inhibited EGFR-LTC kinase (IC50 0.03 and 0.02 μmol·L<sup>-1</sup>, respectively), further underscoring the impact of NO donors in enhancing the cytotoxicity of this triterpene. However, altering the OAc-3 of 8f to another ester or benzyl group (8h-8l) resulted in the loss of the inhibitory activity of the kinase, highlighting the importance of both the structure of triterpenes and the existence of an NO donor in inhibiting EGFR-LTC kinase [37].

In view of the anticancer effect of high-level NO, several OA derivatives have shown significant anticancer activities [38-42]. In this vein, ZHANG's research group made structural modifications to OA's C-3 position to develop new NO-releasing OA derivatives with superior antitumor activity to

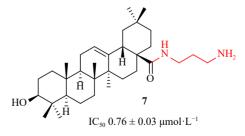


Fig. 8 Structures of OA derivative 7.

OA [43]. These derivatives were designed to be activated by cytochrome P450 (CYP) enzymes, which are differentially expressed in normal and tumor hepatocytes. Among these derivatives, compounds **9a**–**9f** (Fig. 10) released more NO in SMMC-7721 cells, and their inhibitory effects on the proliferation of SMMC-7721 and HepG2 cells were stronger than those of OA and other compounds. Compound **9c** with the highest activity showed potent anti-hepatoma activity while exerting minimal effects on normal hepatocytes. The addition of an NO scavenger significantly reduced the inhibitory activity of **9c** on cancer cells, suggesting that NO might contribute to the antitumor activity of **9c**.

Several derivatives of pentacyclic triterpenoids have been recognized for their significant antitumor activities both in vitro and in vivo [44-47]. Given that the mitochondrial function of cancer cells is different from that of normal cells [44], targeting mitochondria has become a strategy for cancer therapy [45, 46]. Several derivatives of pentacyclic triterpenoids [47] showed significant antitumor activity in vitro and in vivo. In particular, triterpenes featuring a piperazinyl spacer and a rhodamine B moiety have been identified as mitogen agents with high cytotoxicity [48-52]. SRB analysis unveiled that all these triterpene homopiperazine-rhodamine adducts exhibited high cytotoxicity to a range of human tumor cell lines, but their cytotoxicity to non-malignant cells was significantly lower, and these compounds were even more cytotoxic than previously prepared piperazinyl analogs [53]. Compound 10 (Fig. 11) stood out as the most cytotoxic in the series.

OA and cinnamic acid (CA) are two natural products with anticancer effects. CA has demonstrated inhibitory effects on various cancer types, including lung cancer, prostate cancer, melanoma, and glioblastoma <sup>[54-56]</sup>. In order to improve the anticancer activity of OA, WANG *et al.* designed and synthesized a series of new OA-CA esters by a molecular hybridization strategy <sup>[57]</sup>. The cytotoxicities of these newly synthesized compounds to HeLa, MCF-7, and L-O2

cell lines were assessed *in vitro* using the MTT method. Among these compounds, **11a** (Fig. 12) exhibited the highest selectivity and cytotoxicity to HeLa cells (IC<sub>50</sub> 1.35  $\mu$ mol·L<sup>-1</sup>) but had no inhibitory activity on MCF-7 cells (IC<sub>50</sub> > 100  $\mu$ mol·L<sup>-1</sup>) and L-O2 cells (IC<sub>50</sub> > 100  $\mu$ mol·L<sup>-1</sup>). Conversely, **11b** displayed the most potent selective inhibition on MCF-7 cells (IC<sub>50</sub> 1.79  $\mu$ mol·L<sup>-1</sup>).

CHENG's team reported several pharmacophore-based OA derivatives (Fig. 13) that were coupled with antitumor drug 5-FU to evaluate their antitumor activities  $^{[58]}$ . The results revealed that monosubstituted hybrids 12a and 12b possessed notable antiproliferative activities, while disubstituted hybrids 12c and 12d displayed no antitumor activities. Hybrid 12b showed potential selectivity to tumor cells (K562, IC<sub>50</sub> 22.99 µmol·L<sup>-1</sup>) and moderate antiproliferative activities against MDR cell lines A549/T (IC<sub>50</sub> 43.07 µmol·L<sup>-1</sup>) and Bel-7402/FU (IC<sub>50</sub> 31.42 µmol·L<sup>-1</sup>), which were comparable to its effect on A549 (IC<sub>50</sub> 50.54 µmol·L<sup>-1</sup>) and Bel-7402 cell lines (IC<sub>50</sub> 43.82 µmol·L<sup>-1</sup>).

#### Antiviral activity of OA and its derivatives

Influenza A (IAV) and B (IBV) viruses are significant contributors to upper respiratory tract infections [59]. Pentacyclic triterpenes and their derivatives from plants showed inhibitory activity on the entry of IAV virus in vitro. In particular, compound 13a had strong anti-H1N1 entry activity with an IC<sub>50</sub> of 4.05  $\mu$ mol·L<sup>-1</sup> [59]. OA has emerged as a promising scaffold for developing new influenza virus entry inhibitors [60]. It is reported that the anti-influenza activity of OA can be markedly improved by conjugating its 17-COOH or 3-OH groups with oligosaccharides [59, 61-63]. Expanding on this concept, SU et al. conjugated different saccharide moieties with 28-COOH of OA using a triazole linker, assessing the anti-influenza activity of these compounds in vitro [60]. Among them, compound 13b (Fig. 14), an OA-glucose conjugate, demonstrated the strongest inhibitory activity against influenza A/WSN/33 (H1N1) virus with an IC<sub>50</sub> value of 5.47

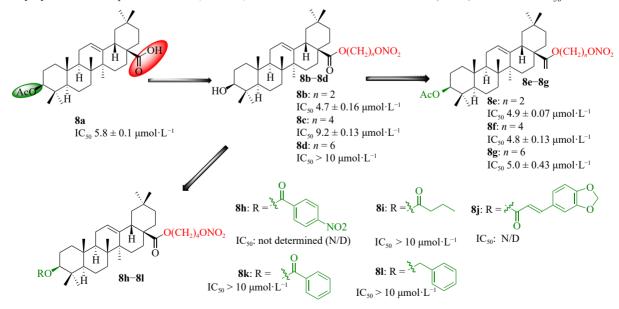


Fig. 9 Structures of OA derivatives 8a-81.

9a: 
$$R_1 = \frac{1}{2} \times N_1$$
 $R_2 = Glu$ 
 $IC_{50} 10.52 \pm 0.91 \, \mu mol \cdot L^{-1}$ 

9b:  $R_1 = \frac{1}{2} \times N_1$ 
 $R_2 = Gal(OAc)_4$ 
 $R_3 = \frac{1}{2} \times N_2$ 
 $R_4 = Ph$ 
 $R_4 = Ph$ 
 $R_4 = Bh$ 
 $R_4 = Bh$ 
 $R_4 = Bh$ 
 $R_5 = Gal(OAc)_4$ 
 $R_1 = \frac{1}{2} \times N_2$ 
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 $R_7 = \frac{1}{2} \times N_2$ 
 $R_8 = \frac{1}{2} \times N_2$ 
 $R$ 

Fig. 10 Structures of OA derivatives 9a-9f.

$$R_{1}$$

$$R_{1} = NO_{2}, R_{2} = CH_{3}$$

$$R_{2} = CH_{3}$$

$$R_{1} = NO_{2} = CH_{3}$$

$$R_{2} = CH_{3}$$

$$R_{2} = CH_{3}$$

$$R_{3} = NO_{2} = CH_{3}$$

$$R_{2} = CH_{3}$$

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$$R_{3} = NO_{2} = CH_{3}$$

$$R_{3} = NO_{2} = CH_{3}$$

$$R_{4} = NO_{2} = CH_{3}$$

$$R_{5} = CH_{3} =$$

Fig. 11 Structures of OA derivative 10.

μmol· $L^{-1}$  and showed no obvious cytotoxic effect on MDCK (Madin-Darby canine kidney) cells at 100 μmol· $L^{-1}$ . Further broad-spectrum anti-influenza experiments revealed that **13b** was potent against five different strains, including IAV and IBV, with IC<sub>50</sub> values in the low micromolar range. Hemagglutination inhibition (HI) assays and docking experiments indicated that **13b** might interfere with influenza virus infection by targeting the hemagglutinin (HA) protein.

The development of hepatitis C virus (HCV) entry inhibitors represents an emerging approach that satisfies a tandem mechanism for use with other inhibitors in a multifaceted cocktail. Screening of Chinese herbal extracts revealed that OA displayed a weak capability to inhibit HCV entry with an IC<sub>50</sub> value of 10  $\mu$ mol·L<sup>-1</sup> [<sup>64</sup>]. Chemical exploration of this triterpene compound revealed its pharmacophore requirements for blocking HCV entry, with rings A, B, and E being conserved while ring C and ring D being tolerant to some

Fig. 12 Structures of OA derivatives 11a and 11b.

modifications [64]. Hydroxylation at 16-C (14a, Fig. 15) significantly enhanced its potency for inhibiting HCV entry, with an IC<sub>50</sub> value of 1.4 μmol·L<sup>-1</sup>. Further modification by conjugation of this new lead with a disaccharide at 28-COOH removed the undesired hemolytic effect and, more importantly, increased its potency by about 5-fold (IC<sub>50</sub> 0.3 µmol·L<sup>-1</sup>) (14b, Fig. 15). Alternatively, the combination of two pharmacophores is a potential strategy to increase the activity of 14a. Unexpectedly, 14c, the 14a dimer bearing a triazole linker, exhibited significantly higher potency with an IC50 value of 10.3 nmol·L<sup>-1</sup>, almost two orders of magnitude more potent than 14a (IC<sub>50</sub> 1.4  $\mu$ mol·L<sup>-1</sup>). 14c was the most potent compound based on the HCVpp entry assay. Mechanistically, these functional triterpenes interrupt the interaction between the HCV envelope protein E2 and its receptor CD81 by binding to E2, thus blocking the recognition process between the virus and the host cell.

YANG et al. have made significant strides in the devel-

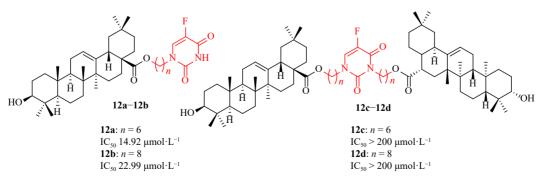


Fig. 13 Structures of OA derivatives 12a-12d.

opment of novel antiviral strategies with their work on multivalent OA protein conjugates, which function as non-glycosylated mucin mimics for the capture and entry inhibition of influenza viruses [65]. They synthesized an OA derivative, compound 15 (Fig. 16), featuring an amine terminal group, by esterification of carboxylic acid and then grafted it onto human serum albumin (HSA) using the diethyl squarate strategy. The resulting conjugate exhibited remarkable agglutination properties, strong capture efficiency, and a high-affinity constant for viral particles. These attributes suggest that the protein conjugate may be used as an anti-infection barrier to prevent the virus from invading host cells.

Medina-O'Donnell et al. evaluated how the derivatives of OA conjugated with one or two amino acids and an acyl group affect the inhibition of HIV-1 protease [66]. Their in vitro studies suggested that the inclusion of a carboxyacyl group generally enhanced the inhibition of HIV-1 protease, especially when a phthaloyl group was present, with IC50 values below 5 µmol·L<sup>-1</sup>. Among these derivatives, three 3-

Fig. 14 Structures of OA derivatives 13a and 13b.

phthaloyl OA compounds (16, Fig. 17) displayed submicromolar IC<sub>50</sub> values, indicating an activity level 60- and 100fold greater than OA. These findings suggest that the conjugation of OA with one or two amino acids and a phthalovl group improves its efficacy as an HIV-1 protease inhibitor, pointing to these triterpene derivatives as potential antiviral agents against HIV. The study also yielded several key insights into the mechanisms of OA derivatives in inhibiting HIV-1 protease. For instance, it was observed that coupling short-chain  $\omega$ -amino acids ( $\gamma$ -aminobutyric acid or 6AHA) to the scaffold (OA) at the C-28 position enhances the inhibitory effect on HIV1 protease. Conversely, the presence of long-chain omega-amino acids, like 11AUA, attenuates this effect. The results of docking studies showed that the strong inhibitory activity of 3-phthaloyl OA derivatives may be attributed to their ability to form more hydrogen bonds, which facilitates the placement of ligands in a more favorable position in the proteases. However, the presence of long-chain amino acids may prevent ligands from establishing favorable contacts with the active sites of the enzyme, resulting in a loss of activity unless phthaloyl groups are present.

LI et al. synthesized a series of pentacyclic triterpene derivatives modified by coupling with various polyphenols on C-28 and evaluated the antiviral activity of M against influenza A H1N1 virus in MDCK cells [67]. Three of these compounds 17a, 17b, and 17c (Fig. 18) showed strong anti-influenza efficacy, surpassing the standard antiviral drug oseltamivir. Importantly, these compounds exhibited no toxicity to MDCK cells at a concentration of 100 µmol·L<sup>-1</sup>. Compound 17c, in particular, is one of the most representative conjugates in this series. It showed the strongest inhibitory activity against influenza A/WSN/33 (H1N1) viruses with an  $IC_{50}$  value of 5.80 umol·L<sup>-1</sup> and a stronger affinity with HA with a KD value of 15.6 µmol·L<sup>-1</sup>. Further investigations, including hemagglutination inhibition (HI) assays, surface plasmon resonance, and molecular simulations, indicated that showed that these conjugates bind closely to the viral envelope HA (K<sub>d</sub> 15.6 μmol·L<sup>-1</sup>) to prevent the influenza virus from invading host cells.

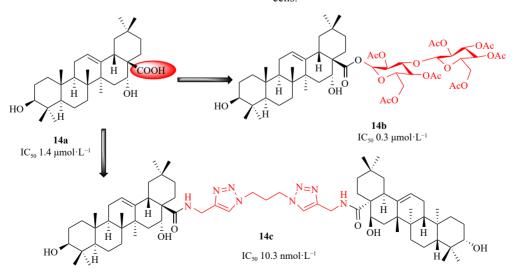


Fig. 15 Structures of OA derivatives 14a-14c.



Fig. 16 Structures of OA derivative 15.

R: phthaloyl group Aa:  $\omega$ -amino acid or  $\omega$ -and  $\alpha$ -amino acid IC<sub>50</sub> from  $0.79 \pm 0.01$  to  $3.1 \pm 0.10$   $\mu$ mol·L<sup>-1</sup>

Fig. 17 Structures of OA derivative 16.

$$\begin{array}{c} R_{2} \\ R_{3} \\ R_{4} \\ R_{4} \\ R_{5} \\$$

Fig. 18 Structures of OA derivatives 17a-17c.

Inspired by the discovery of a series of pentacyclic triterpenes targeting influenza virus HA protein as entry inhibitors, MENG *et al.* designed and synthesized a series of OA analogs by linking different amino acids to the 28-COOH of OA <sup>[68]</sup>. The antiviral activities of these compounds were evaluated *in vitro*. Among these newly synthesized compounds, **18** (Fig. 19) showed significant antiviral activity against influenza A/WSN/33 (H1N1) viruses. HI assays and docking experiments provided insights into the mechanism of action of these OA analogs. It was observed that they might block the interaction between the influenza virus HA protein and host cell sialic acid receptor by binding to the HA protein, thus blocking virus invasion and producing the same effect as parent compounds.

#### Hypoglycemic activity of OA and its derivatives

 $\alpha$ -Glucosidase inhibitors are a class of oral hypoglycemic drugs used for the treatment of diabetes that delay the absorption of intestinal carbohydrates.  $\alpha$ -Glucosidase inhibitors are relatively mature drugs for the treatment of diabetes and have been widely used in the clinic. ZHONG *et al.* synthesized a series of novel anti- $\alpha$ -glucosidase OA analogs and evaluated their biological activities *in vitro* and *in vivo*, as shown in Fig. 20 <sup>[69]</sup>. The results of  $\alpha$ -glucosidase inhibitory activity *in vitro* showed that all the designed compounds had obvious inhibitory activity. The IC<sub>50</sub> values of compounds 19a, 19b, 19c, and 19d were 0.33 ± 0.01, 0.98 ± 0.06, 0.69 ±

Fig. 19 Structures of OA derivative 18.

0.01, and  $0.72 \pm 0.21~\mu mol \cdot L^{-1}$ , respectively. The study of enzyme kinetics indicated that derivatives 19a-19d functioned as non-competitive inhibitors of  $\alpha$ -glucosidase. Molecular docking studies further revealed that these four compounds could interact with the hydrophobic region of the active pocket to form hydrogen bonds, thus enhancing their binding affinity to  $\alpha$ -glucosidase. Cytotoxicity tests confirmed that the active compounds 19a-19d exhibited no cytotoxicity to normal 3T3 cell lines. The study on the actual pharmacological potential of derivative 19a-19d in vivo showed that their hypoglycemic effect was similar to that of acarbose, a positive control.

Protein tyrosine phosphatase 1B (PTP1B) is a negative regulatory factor in insulin signal transduction and an effective drug target for the treatment of diabetes and obesity. QI-AN *et al.* synthesized a series of oleic acid derivatives and evaluated their inhibitory activity against PTP1B [<sup>70]</sup>. Among them, compound **20** emerged as the most effective in inhibiting PTP1B (IC<sub>50</sub> 3.12 μmol·L<sup>-1</sup>). The structure of compound **20** is shown in Fig. 21. The SAR analysis of these derivatives unveiled that the integrity of the A ring and 12-*ene* part was crucial for maintaining the enzyme inhibitory activity of PTP1B. Furthermore, the distance between the hydrophilic group and the acidic group and between oleanolene and the acidic group played a role in determining the PTP1B inhibitory activity.

Studies have shown that PTP-1B enzymes that deactivate insulin and leptin receptors are involved in glucose and fat metabolism, respectively. Pentacyclic acid triterpenes, such as oleic acid, have been recognized as a class of potent PTP-1B inhibitors. Studies have indicated that the inhibitory activity of these compounds can be significantly enhanced through the substitution of the C-3 hydroxyl, C-12 double bond, or C-28 carboxylic acid groups with aryl or alkyl groups [71]. Based on these studies, Ramírez-Espinosa et al. synthesized a series of OA derivatives [72], as shown in Fig. 22. The study highlighted that the presence of carboxylic acid on C-28 and/or its corresponding reduction product methanol derivative (hydrogen bond donor) is essential for maintaining of these compounds against PTP-1B. Moreover, the formation of esters or ethers on C-3 further enhanced this inhibitory activity. In the non-insulin dependent diabetic rat model, OA derivatives showed an enhanced inhibitory effect on PTP-1B activity by increasing their molecular interaction with catalytic or allosteric sites and hypoglycemic effect.

#### Hypolipidemic activity of OA and its derivatives

Cholesteryl ester transfer protein (CETP) is a key thera-



$$\begin{array}{c} \textbf{19a: R} = -\frac{1}{2} & \textbf{19b: R} = -\frac{1}{2} \\ \textbf{IC}_{50} \ 0.33 \pm 0.01 \ \mu \text{mol} \cdot \textbf{L}^{-1} \\ \textbf{19d: R} = -\frac{1}{2} & \textbf{IC}_{50} \ 0.98 \pm 0.06 \ \mu \text{mol} \cdot \textbf{L}^{-1} \\ \textbf{19d: R} = -\frac{1}{2} & \textbf{IC}_{50} \ 0.72 \pm 0.21 \ \mu \text{mol} \cdot \textbf{L}^{-1} \\ \textbf{IC}_{50} \ 0.69 \pm 0.01 \ \mu \text{mol} \cdot \textbf{L}^{-1} \\ \textbf{IC}_{50} \ 0.72 \pm 0.21 \ \mu \text{mol} \cdot \textbf{L}^{-1} \\ \end{array}$$

Fig. 20 Structures of OA derivatives 19a-19d.

Anti-inflammatory activity of OA and its derivatives

Arachidonic acid (AA) plays a critical role in the inflammatory process. It is released from cell membrane phospholipids through the action of phospholipase A2 in response to inflammation-related cell injury. Three key enzymes control

the AA pathway: cyclooxygenases (COX-1 and COX-2),

which produce prostaglandins (PG) and thromboxane; lipoxy-

genases (LOX, including 5-LOX and 15-LOX), which gener-

ate leukotrienes (LTS) and hydroxy eicosatetraenoic acids

(HETE); and cytochrome P450 enzymes (CYPS), which produce epoxide eicosatrienoic acids [74-78]. The products of COX

and LOX pathways are key bioactive lipid mediators in the

induction of pathophysiological inflammatory conditions re-

quiring drug intervention. In 2019, Vo et al. investigated the

inhibitory activities of 29 natural oleanane and ursolane

pentacyclic triterpenes on four main enzymes involved in the

inflammatory process: 5-LOX, 15-LOX-2, COX-1, and COX-

2. One of these compounds, 3-O-acetyl- $\beta$ -boswellic acid (23,

Fig. 24), exhibited an obvious inhibitory effect on human 5-

LOX-2 (IC<sub>50</sub> 12.2  $\pm$  0.47  $\mu$ mol·L<sup>-1</sup>) <sup>[79]</sup>. The SAR analysis re-

vealed that the existence of 24 hydroxyl groups was benefi-

cial to the inhibition of 5-LOX and COX-1. The introduction

of a carboxylic acid group at position 30 was crucial for the

dual inhibitory activity of 5-LOX/COX, and the carbonyl

group binding to Cmur11 significantly enhanced the inhibit-

ory activity of 5-LOX. The activity of 5-LOX was markedly

Fig. 21 Structures of OA derivative 20.

peutic target for treating atherosclerotic cardiovascular diseases. Molecular simulation studies have revealed that pentacyclic triterpenoids can simulate the interaction between proteins and ligands by occupying the binding site of the endogenous ligand cholesterol ester (CE). Based on these insights, CHANG *et al.* designed and synthesized a series of new pentacyclic triterpenoid CETP inhibitors <sup>[73]</sup>. As shown in Fig. 23, compound **22b** (IC<sub>50</sub> 0.28  $\pm$  0.068  $\mu$ mol·L<sup>-1</sup>) stood out with the best biological activity, which validates the authors' molecular design strategy. Molecular dynamics simulations showed that compared with compound **22a** (IC<sub>50</sub> > 50  $\mu$ mol·L<sup>-1</sup>), compound **22b** showed a more stable hydrogen bond interaction with Ser191 and stronger hydrophobic interactions with Val198 and Phe463, which is the main reason for their different CETP inhibitory activity.

21a: 
$$R_1 = 0$$
 $R_2 = 0$ 
 $R_2 = 0$ 
 $R_3 = 0$ 
 $R_4 = 0$ 
 $R_5 = 0$ 

Fig. 22 Structures of OA derivatives 21a-21k.

$$\begin{array}{c} R_{2} \\ R_{1} \\ R_{2} \\ R_{1} \\ R_{2} \\ R_{1} \\ R_{2} \\ R_{2} \\ R_{2} \\ R_{1} \\ R_{2} \\ R_{2} \\ R_{3} \\ R_{2} \\ R_{3} \\ R_{2} \\ R_{3} \\ R_{2} \\ R_{3} \\$$

Fig. 23 Structures of OA derivatives 22a and 22b.

inhibited by  $\alpha$ -hydroxy at C-2 or a carboxylic acid group at C-23. The experimental results uncovered that the types and configurations of polar groups at positions C-2, -3, -11, -24, and -30 were important structural factors in pentacyclic triterpenes due to their potential for anti-inflammatory lead compounds.

AMPK exists in the form of a heterotrimeric complex ( $\alpha$ -,  $\beta$ -,  $\gamma$ -), which is a key kinase that regulates energy metabolism and stress responses <sup>[80]</sup>. The activation of AMPK can protect the body from tissue damage caused by acute and chronic inflammation <sup>[81]</sup>. Importantly, existing AMPK agonists can play an anti-inflammatory role in a variety of inflammatory models <sup>[82-86]</sup>. It is reported that OA is an activator of AMPK <sup>[87-90]</sup>. LIU *et al.* found that a variety of natural pentacyclic triterpene saponins and saponins could stimulate the phosphorylation of AMPK and identified  $\delta$ -OA (24a) as a powerful AMPK activator <sup>[91]</sup>. With 24a as the lead com-

pound, a series of  $\delta$ -OA saponin derivatives were synthesized to discover more effective and anti-inflammatory drugs with pharmacokinetics. The results of cell experiments denoted that saponin **24b** (Fig. 25) could significantly inhibit the secretion of pro-inflammatory cytokines TNF- $\alpha$  and IL-6 stimulated by endotoxin in macrophages. The preliminary mechanism showed that **24b** could stimulate the phosphorylation of AMPK and acetyl-CoA carboxylase (ACC). Compared with that of aglycone, the bioavailability of **24b** was substantially improved. More importantly, **24b** exhibited obvious anti-inflammatory and hepatoprotective effects in a mouse model of fulminant liver failure induced by LPS/D-GalN.

In order to explore the effects of different substituents on anti-inflammatory activity, Bhandari et al. synthesized analogs (25a-25m, Fig. 26) by modifying the structure of OA at the C-3 and C-28 positions. They evaluated their anti-inflammatory effects on the production of NO in macrophages induced by lipopolysaccharide (LPS) [92]. The results revealed that compared with NO synthase inhibitor L-NAME (RAW 264.7 cells,  $IC_{50}$  69.21 ± 2.65 µmol·L<sup>-1</sup>; J774A.1 cells,  $IC_{50}$  $73.18 \pm 1.70 \,\mu\text{mol} \cdot \text{L}^{-1}$ ), all synthetic OA analogs could inhibit NO production. Specifically, compounds 25a, 25g, 25h, and 25i demonstrated a strong ability to inhibit NO production. Among them, 25i (IC<sub>50</sub> of RAW 264.7 and J774A.1 cells were  $2.66 \pm 1.54$  and  $10.8 \pm 2.31 \,\mu\text{mol}\cdot\text{L}^{-1}$ , respectively) inhibited NO production by 20 times as much as the parent compound, without affecting cell viability. The most potent NO inhibitors (25a, 25g, 25h, and 25i) slightly inhib-

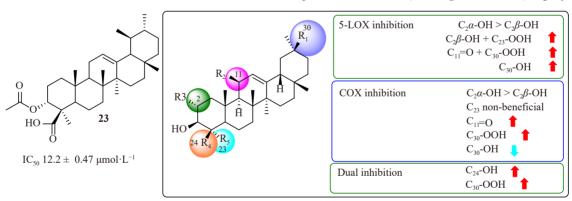


Fig. 24 Structures of OA derivatives 23. 2, 11, 23, 24, and 30 marked in the figure are the labels of the C atoms.  $R_1$ – $R_5$  refer to the groups connected to the C atoms, and the specific groups have been marked in this figure.

Fig. 25 Structures of OA derivatives 24a-24b.

25b: 
$$R_1 = O$$
,  $R_2 = H$ 
 $IC_{50} 13.8 \pm 1.34 \, \mu mol \cdot L^{-1}$ 
25c:  $R_3 = H$ 
 $IC_{50} 13.8 \pm 2.76 \, \mu mol \cdot L^{-1}$ 
25g:  $R_3 = CI$ 
 $IC_{50} 12.8 \pm 2.34 \, \mu mol \cdot L^{-1}$ 
25g:  $R_3 = CI$ 
 $IC_{50} 12.8 \pm 2.34 \, \mu mol \cdot L^{-1}$ 
25g:  $R_3 = CI$ 
 $IC_{50} 12.8 \pm 2.34 \, \mu mol \cdot L^{-1}$ 
25g:  $R_3 = CI$ 
 $IC_{50} 12.8 \pm 2.34 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 2.75 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 2.17 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 2.17 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 2.17 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 2.17 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 
25h:  $R_3 = -\frac{1}{8} \, N - CH_3$ 
 $IC_{50} 12.8 \pm 1.38 \, \mu mol \cdot L^{-1}$ 

Fig. 26 Structures of OA derivatives 25a-25m.

ited the LPS-induced production of TNF- $\alpha$  (27.9%–51.9%) but had no inhibitory effect on the production of IL-1 $\beta$  (11.1%–37.5%) [92].

In 2021, Jannus et al. used an inflammatory model to study the anti-inflammatory effects of diamine-polyethylene glycol derivatives of OA (26, Fig. 27) in vitro and in vivo [93]. They tested the anti-inflammatory activity of compound 26 in RAW264.7 cells. At the same time, the production of NO was monitored. When the concentration of derivative 26 was 11 μg·mL<sup>-1</sup>, the inhibition rate of NO was more than 75%. Cell cycle analysis showed that the LPS-induced G<sub>0</sub>/G<sub>1</sub> phase arrest of RAW264.7 cells was reversed. In addition, through Western blotting analysis, they identified the molecular mechanism by which derivative 26 may be activated: inhibiting the expression of cytokines such as TNF- $\alpha$ , IL-1β, iNOS, and COX-2 and blocking the production of RAW264.7 p-IκBα stimulated by endotoxin. Furthermore, Jannus et al. investigated the in vivo anti-inflammatory effects of derivatives 26 and tetradecyl phorbol ester (TPA) in the treatment of acute ear swelling in male BL/6J mice. Compared with diclofenac, derivative 26 had a greater inhibitory effect on swelling and reduced the thickness of the ear by

#### Hepatoprotective activity of OA and its derivatives

OA has been used as a dietary supplement and is available over-the-counter as a treatment for liver diseases in humans. YU *et al.* designed and synthesized two new OA prodrugs, **27a** and **27b**, using the slow-release properties of 1,3-cyclopropyl phosphate to modify OA metabolism, as shown in Fig. 28 [94]. The results showed that these prodrugs had strong antioxidant activities and obvious protective effects

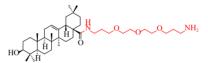


Fig. 27 Structures of OA derivative 26.

against liver injury caused by CCl<sub>4</sub>. In terms of metabolism, the study revealed that the half-life of the prodrugs was extended, and their pharmacokinetic parameters significantly changed after direct administration. Further biodistribution studies demonstrated that the concentrations of prodrugs and OA were the highest in the liver, and most of them were excreted *via* the stool. The findings suggest that the incorporation of 1,3-cyclopropyl phosphate into OA to form prodrugs represents an innovative approach to enhance the efficacy and safety profile of the parent drug.

#### Antibacterial activity of OA and its derivatives

Many compounds with guanidine have significant antibacterial and antifungal activities <sup>[95-97]</sup>. At physiological pH, the guanidine group is positively charged. One of the antibacterial mechanisms of guanidine ion salts may be the electrostatic interaction between the negatively charged bacterial surface and the positively charged compound, potentially damaging bacterial cells. YU *et al.* synthesized guanidine functionalized OA derivatives in 2019 <sup>[98]</sup>. The structures of some compounds are shown in Fig. 29, and their antibacterial properties were studied. Compared with the parent molecule, most of the amino and guanidine derivatives of OA showed significantly increased inhibitory activity against methicillin-resistant *Staphylococcus aureus* (MRSA).

A set of OA amide derivatives (Fig. 30) was synthesized

27b (S)

Fig. 28 Structures of OA derivatives 27a and 27b.

by Blanco-Cabra *et al.* <sup>[99]</sup>. One of these derivatives, **29b**, increased the antibacterial activity of the parent compound *in vitro* while reducing their toxicity to most Gram-positive bacteria, including MRSA. It is noteworthy that OA derivative **29b** increased its antimicrobial activity and reduced the MIC<sub>50</sub> against MRSA by 87% (MIC<sub>50</sub> of 10 μg·mL<sup>-1</sup>) compared with the parent compound (OA, MIC<sub>50</sub> 75 μg·mL<sup>-1</sup>). Similarly, another derivative, **29c**, also demonstrated a reduction in MRSA MIC<sub>50</sub> by 60%. However, bimolecular OA de-

rivatives **29e–29g** linked *via* diamines generally lacked anti-bacterial activity.

In 2020, ZHOU et al. reported the effects of OA and its analogs on the  $\beta$ -lactamases (NDM-1, KPC-2, and VIM-1) of Enterobacteriaceae and  $\beta$ -endocrine of Staphylococcus aureus [100]. The activity of amidase ( $\beta$ -lactamase N1) has a significant inhibitory effect. In the mouse infection model, the efficacy of combining OA and  $\beta$ -lactam drugs was evaluated. The results indicated a notable synergistic effect between the two. The survival rate of mice infected with Staphylococcus aureus and Escherichia coli increased from 25.0% to 75.0% after receiving this combination therapy. In cases where only  $\beta$ -lactam drugs were used, the survival rates were 44.4% and 61.1%, while the addition of OA increased these rates to 77.8%. The study's findings suggest that the combination therapy can simultaneously target drug-resistant enzymes and toxins, which can be a promising treatment strategy for drugresistant bacterial infections.

#### Anti-AD activity of OA and its derivatives

The typical symptoms of Alzheimer's disease (AD), such as amnesia or behavioral disorders, are often attributed to a reduction in acetylcholine (ACh) concentration [101-104]. Typically, the breakdown of this neurotransmitter is mediated by acetylcholinesterase (AChE). Butyrylcholinesterase (BuChE) is also believed to play a crucial role in regulating the concentration of ACh in different tissues. It is speculated that BuChE may compensate for the decrease in AChE activity [105, 106]. Therefore, these two enzymes are potential targets for the treatment of AD or at least a tool for achieving a deep-

$$28a: R = \sqrt[3]{5}$$

$$28b: R = \sqrt[3]{5}$$

$$NH_{2}$$

$$28b: R = \sqrt[3]{5}$$

$$NH_{2}$$

$$28c: R = \sqrt[3]{5}$$

$$NH_{2}$$

$$28f: R = \sqrt[3]{5}$$

$$NH_{2}$$

$$NH_{2}$$

$$NH_{2}$$

$$NH_{2}$$

Fig. 29 Structures of OA derivatives **28a-28f** (**28b-28f**, MICs  $\leq$  0.25  $\mu$ g·mL<sup>-1</sup>).

HO 
$$\frac{H}{\tilde{H}}$$
  $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{H}{\tilde{H}}$   $\frac{1}{\tilde{H}}$   $\frac{1}{\tilde{H}}$ 

29a: n = 1 (No activity)
29b: n = 4 MIC<sub>50</sub> = 10  $\mu$ g·mL<sup>-1</sup> (against MRSA)
29f: n = 8 MIC<sub>50</sub> = 30  $\mu$ g·mL<sup>-1</sup> (against MRSA)
29g: n = 8 MIC<sub>50</sub> = 30  $\mu$ g·mL<sup>-1</sup> (against MRSA)
29g: n = 8 (No activity)
29d: n = 1, the terminal amine group is NMe, (No activity)

Fig. 30 Structures of OA derivatives 29a-29g.

er understanding of AD pathology.

Loesche *et al.* synthesized several OA derivatives, evaluated their inhibitory effects on AChE and BuChE by the Ellman method, and determined the type of inhibition of each active compound [107]. The results showed that several compounds (**30a**–**30e**, as shown in Fig. 31) exerted a stronger inhibitory effect on AChE than OA, which acted as a mixed inhibitor of AChE. Their inhibition constants were  $K_i$  3.46  $\pm$  0.56, 6.37  $\pm$  0.26, 4.22  $\pm$  0.68, and 6.26  $\pm$  2.71  $\mu$ mol·L<sup>-1</sup>. Compound **30a** demonstrated the highest activity, which was

three times that of OA. Most of the OA derivatives had no inhibitory effect on BuChE, except for compound **30e** ( $K_i$  24.35  $\pm$  9.07  $\mu$ mol·L<sup>-1</sup>).

In 2018, according to previous reports on oleanolic-glycoside saponins <sup>[108, 109]</sup>, LI *et al.* in our research group synthesized hederacolochiside E and a series of new derivatives **31a–31f** (Fig. 32) in a simple and practical way for the first time <sup>[110]</sup>. The neuroprotective effects of these compounds on injuries induced by  $H_2O_2$  and  $A\beta_{1.42}$  were further evaluated. Compound **31b**, a relatively simple chemical structure, was

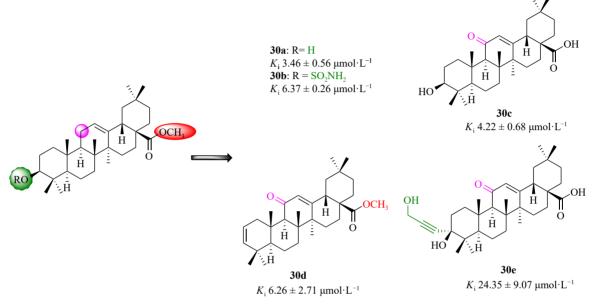


Fig. 31 Structures of OA derivatives 30a-30e.

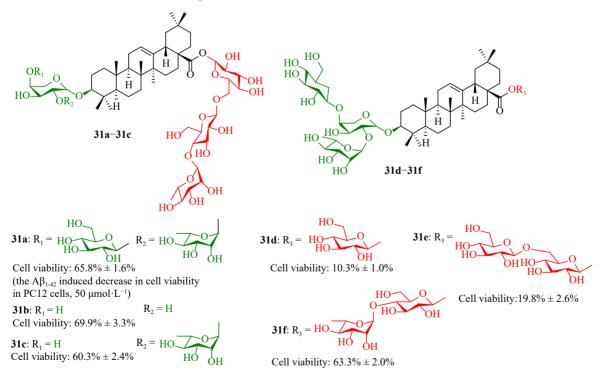


Fig. 32 Structures of OA derivative 31a-31f.

the most active derivative, which exhibited a better neuroprotective effect in the  $A\beta_{1-42}$ -induced injury model. Moreover, 31a and 31b remarkably reduced the release of lactate dehydrogenase (LDH), intracellular ROS level, and the extent of malondialdehyde (MDA) increase resulting from  $A\beta_{1-42}$  treatment, indicating that these compounds showed neuroprotective effects in AD models by regulating oxidative stress. The SAR analysis revealed that the rhamnopyranosyl residue at the C-28 position was of vital importance for the neuroprotective activity.

Accumulating evidence suggests that neurotoxic amyloidbeta (AB) peptides are one of the major causative factors of AD, particularly  $A\beta_{42}$ , a major contributing factor to AD pathogenesis.  $A\beta_{42}$  is known for its propensity to aggregate, and its accumulation is considered a key biomarker and likely a primary driver of AD [111]. In 2020, LUO et al. evaluated the anti-AD biological activity of benzyl-OA carbamate derivatives [112]. They selected compound 32 (Fig. 33) as carbamate compounds, which are nonnative substrates of endogenous lipase enzymes and are cleaved less efficiently than an ester in vivo. They found that compound 32 reduced the production of  $A\beta_{40}$  and  $A\beta_{42}$ , with  $IC_{50}$  values of 0.57 and 0.65 μmol·L<sup>-1</sup>, respectively. The treatment with compound 32 reduced the  $A\beta$  level in N2a695 cell-conditioned media by 10%-40%. Furthermore, **32** and OA at 5 μmol·L<sup>-1</sup> exhibited no toxicity against N2a695 cells in an MTT assay.

BnHN 
$$\frac{1}{2}$$
  $\frac{1}{2}$   $\frac{1}{2}$ 

Fig. 33 Structures of OA derivative 32.

#### Conclusion

The utilization of natural products as lead compounds for drug discovery is a well-established and significant aspect of new drug development. OA, a natural product of pentacyclic triterpenes widely existing in nature, is a prime example of this approach. Its unique skeleton and specific chiral methyl configuration are the results of long-term natural evolutionary selection, potentially conferring a higher specificity towards biological cell target proteins. Based on the pharmacological effects of OA, such as antitumor, antiviral, antibacterial, anti-inflammatory, hepatoprotective, hypoglycemia, hypolipidemia, and anti-AD properties, this paper outlines the modification strategies and SARs of OA and its derivatives, drawing on two decades of research, in order to provide a valuable reference for future studies on OA. We can be optimistic that further progress will be made in future research on OA and its derivatives, freeing humans from psychosomatic diseases.

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