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•Original article•

Five new spirosterol saponins from Allii Macrostemonis Bulbus

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[ABSTRACT] Five new spirostanol saponins (1–5) and seven known compounds (6–12) were isolated from the *n*-butanol fraction of 75% ethanol extract of Allii Macrostemonis Bulbus. The identification and structural elucidation of all the isolates were performed through extensive 1D and 2D NMR experiments, HR-ESI-MS data analysis and comparisons with literature values. Antioxidant evaluation showed that compounds 6–11 exhibited certain scavenging effects on ABTS radical, where compounds 6, 7 and 11 had IC₅₀ values of 0.208, 0.057 and 0.014 mg·mL⁻¹, respectively.

[KEY WORDS] Allii Macrostemonis Bulbus; Spirosterol saponin; Antioxidant[CLC Number] R284.1[Document code] A[Article ID] 2095-6975(2023)03-0226-07

Introduction

Allii Macrostemonis Bulbus (Xiebai in Chinese) is the dried bulb of *Allium macrostemon* Bge. or *Allium chinense* G. Don which are perennial herbs of Liliaceae ^[1]. It is an important medicine for the treatment of "chest impediment". Pharmacological studies have revealed that Allii Macrostemonis Bulbus has antiplatelet aggregation, antitumor, anti-atherosclerosis, antioxidant, lipid-lowering, anti-spasmodic, antibacterial, and other bioactivities ^[2]. However, the pharmacodynamic material basis of this drug has not been clarified. Previous studies have reported that Allii Macrostemonis Bulbus contains abundant bioactive components, such as steroidal saponins ^[3-8], volatile oil ^[9-11], polysaccharides ^[12-14], and nitrogen-containing compounds ^[15, 16]. Among them, steroidal saponins are the main constitutes, and many saponins such as macrostemonosides A–S and chinenosides I–VII, have been isolated.

In this paper, continuing investigation of the *n*-butanol fraction of 75% ethanol extract through detailed chemical investigations by chromatography resulted in the isolation and identification of 12 compounds. Among them, compounds 1-5 are new compounds (Fig. 1), and compounds 6-8 are

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isolated from Allii Macrostemonis Bulbus for the first time. Moreover, the antioxidant activities of the isolated compounds were investigated. Herein, the detail of isolation, structural elucidation, and antioxidant evaluation of these compounds were reported.

Results and Discussion

Compound 1 was isolated as white powder. Its reactions to Liebermann-Burchard and Molisch were positive, and to Ehrlich reagent was negative. After the thin layer of compound 1 was formed, the color developed with concentrated $H_2SO_4/EtOH$ solution (1 : 9, V/V) was purplish red, which suggested that the compound may be a spirosterol saponin. The molecular formula C₅₁H₈₀O₂₄ was established from the HR-ESI-MS spectrum with a positive-ion peak at m/z1099.4965 $[M + Na]^+$ (Calcd. 1099.4972). Analysis of its ¹H NMR spectrum (Table 1) revealed the presence of three olefinic protons at $\delta_{H}4.84$ (1H, s, H-27), 4.80 (1H, s, H-27) and 5.19 (1H, m, H-6), three methyl protons at $\delta_{\rm H}0.88$ (3H, s, Me-18), 0.83 (3H, s, Me-19), 1.11 (3H, d, J = 6.6 Hz, Me-21) of aglycone, and four anomeric protons at $\delta_{\rm H}$ 5.62 (1H, d, J = 7.8 Hz, H-1'), 5.30 (1H, d, J = 7.8 Hz, H-1"), 5.34 (1H, d, J = 7.8 Hz, H-1""), 4.92 (1H, d, J = 7.8 Hz, H-1"") of four sugar moieties. The ¹³C NMR data (Table 2) showed a total of 51 carbon signals, where 27 carbons were assigned to the aglycone and the remaining 24 carbons to four sugar moieties. It revealed four terminal carbon signals at $\delta_{\rm C}$ 103.63, 106.07, 105.49, 105.87 of four sugar moieties, while four character-



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Fig. 1 Structures of compounds 1–5

istic alkene carbon signals at δ_{C} 145.33 (C-25), 141.88 (C-5), 122.54 (C-6), 109.69 (C-27) implied double bonds at C-5/C-6 and C-25/C-27. The proton signals were assigned to the corresponding carbons through direct ¹H and ¹³C correlations in the HMQC spectrum. In the HMBC spectrum (Fig. 2), there were the correlations from H-19 ($\delta_{\rm H}0.83$, 3H, s) to C-1 $(\delta_{C}37.93)$, C-10 $(\delta_{C}38.35)$, C-9 $(\delta_{C}42.74)$, C-5 $(\delta_{C}141.88)$, from H-18 ($\delta_{\rm H}$ 0.88, 3H, s) to C-13 ($\delta_{\rm C}$ 41.36), C-14 ($\delta_{\rm C}$ 51.11), C-17 (δ_{C} 57.49), C-12 (δ_{C} 77.75). Meanwhile, the HMBC correlations from H-21 ($\delta_{\rm H}$ 1.11, 3H, d, J = 6.6 Hz) to C-17 (δ_C 57.49), C-20 (δ_C 40.68), C-22 (δ_C 110.36), and from H₂-27 to C-25 (δ_C145.33), C-26 (δ_C65.78), C-24 (δ_C34.09) suggested that compound 1 existed a spirostenoside skeleton. From the ¹H-¹H COSY analysis, the correlations between H₂-1/H₂-2/H-3 and H₂-11/H-12 revealed that δ_{C} 81.19 was attributed to C-3 and C-3 was connected with glycosyl. The HMBC correlations from $\delta_H 4.07$ (1H, m) to C-12 ($\delta_C 77.75$) indicated that the hydroxyl group was attached to C-12, and the hydroxyl group was β -oriented based on the NOESY cross-peak of H-12 with H-17^[5]. Comparison of the ¹³C NMR spectroscopic signals of the aglycone moiety of 1 with literature values ^[3] and an extensive HMQC and HMBC data analysis, showed that the aglycone of 1 was spirostane-5,25(27)-dien- 3β , 12β -diol. Acid hydrolysis of 1 gave D-galactose and Dglucose in a ratio of $1 \div 3$, and the connectivity of the four sugars was mainly based on the HMBC correlations: H-1' $(\delta_{\rm H}5.62, d, J = 7.8 \text{ Hz})$ with C-3 $(\delta_{\rm C}81.19)$ of the aglycone; H-1" ($\delta_{\rm H}$ 5.30, d, J = 7.8 Hz) with C-4' ($\delta_{\rm C}$ 82.34), H-1"' ($\delta_{\rm H}$ 5.34, d, J = 7.8 Hz) with C-3" (δ_{C} 89.40), and H-1"" (δ_{H} 4.92, d, J =7.8 Hz) with C-2" (δ_{C} 79.59), and all the sugars were determined to be β -oriented according to the J values. Comparing the NMR data of the sugar moieties of compound 1 and compound odospiroside ^[17], they were basically consistent, indicated that they had the same sugar chain. Therefore, compound 1 was identified as spirostane-5,25(27)-dien- 3β ,12 β - diol-3-O- β -D-glucopyranosyl(1 \rightarrow 2)[β -D-glucopyranosyl(1 \rightarrow 3)]- β -D-glucopyranosyl(1 \rightarrow 4)- β -D-galactopyranosid.

Compound 2 was obtained as white powder. It has a molecular formula of $C_{51}H_{82}O_{24}$ deduced by HR-ESI-MS {m/z1101.5834 $[M + Na]^+$ (Calcd. 1101.5837). The ¹H NMR spectrum (Table 1) displayed four methyl singlet signals at δ_{H} 0.83 (3H, s, Me-18), $\delta_{\rm H}$ 0.87 (3H, s, Me-19), $\delta_{\rm H}$ 1.15 (3H, d, J = 6.9 Hz, Me-21) and δ 0.69 (3H, s, Me-27) of aglycone. Moremore, the ¹³C NMR spectrum (Table 2) revealed a total of 51 carbon signals, where 27 carbons were assigned to the aglycone and the remaining 24 carbons to the sugar moieties. In fact, the ¹H and ¹³C NMR spectroscopic data of 2 were similar to those of compound 1, except for the absence of one double bond at C-25/C-27 in 2, which was further confirmed by the H-H COSY, HMBC, HMQC and NOESY spectra (Fig. 2). Furthermore, according to the chemical shifts of C-24 (δ_{C} 30.17), C-25 (δ_{C} 31.51) and C-26 (δ_{C} 67.76), C-25 was determined to be β -oriented ^[3]. Therefore, compound 2 was established as 25R-spirostane-5(6)-en-3β,12β-diol-3-O-β-Dglucopyranosyl($1 \rightarrow 2$)-[β -D-glucopyranosyl($1 \rightarrow 3$)]- β -D-glucopyranosyl $(1\rightarrow 4)$ - β -D-galactopyranosid.

Compound **3** was isolated as white powder. Its physicochemical properties were same with those of compounds **1** and **2**. A comparison of the NMR data (Table 2) of **3** and **2** showed that there were no characteristic alkene carbon signals at C-5/C-6 in **3**. The hydrogen signal of Me-18 ($\delta_{\rm H}0.83$) was in the relative low field region compared with Me-19 ($\delta_{\rm H}0.63$), so H-5 was determined to be α -oriented ^[9]. On the basis of the ¹H-¹H COSY, HMBC, HMQC and NOESY spectra (Fig. 2), compound **3** was demonstrated as $25R-5\alpha$ spirostane- 3β , 12β -diol-3-O- β -D-glucopyranosyl($1 \rightarrow 2$)-[β -Dglucopyranosyl($1 \rightarrow 3$)]- β -D-glucopyranosyl($1 \rightarrow 4$)- β -Dgalactopyranosid.

Compound 4 was isolated as white powder and its molecular formula was determined to be $C_{50}H_{82}O_{23}$ based on its

No.	1	2	3	4	5
1	1.68, m; 0.93, m	1.70, m; 0.99, m	1.63, m; 1.00, m	2.20, m; 1.19, m	2.15, m; 1.08, m
2	2.25, m; 2.73, m	1.69, m; 1.37, m	1.34, m; 1.10, m	4.04, m	4.01, m
3	4.56, m	4.71, m	4.57, m	4.37, m	4.52, m
4	2.43, m; 2.66, m	2.42, m; 2.68, m	1.51, m; 0.79, m	1.66, m; 1.39, m	1.27, m; 1.32, m
5	-	-	1.96, t (6.4, 8.6)	1.00, m	1.04, m
6	5.19, m	5.29, m	0.66, m	1.61, m; 1.36, m	1.60, m; 1.31, m
7	1.27, m; 1.50, m	2.09, m; 1.26, m	1.50, m; 1.08, m	2.02, m; 1.44, m	1.74, m; 1.41, m
8	1.66, m	1.67, m	1.65, m	1.37, m	1.39, m
9	2.11, m	1.65, m	2.01, m	0.57, m	0.49, m
10	-	-	-	-	-
11	1.22, m; 0.97, m	1.40, m; 1.81, m	1.03, m; 0.89, m	1.17, m; 1.40, m	1.39, m; 1.18, m
12	4.07, m	3.90, m	4.02, m	1.61, m; 0.77, m	1.42, m; 0.79, m
13	-	-	-	-	-
14	0.95, m	0.63, s	0.49, t (7.3, 14.5)	0.97, m	1.04, t (8.2, 6.8)
15	1.40, m; 1.42, m	1.41, m; 1.51, m	1.66, m; 1.52, m	1.86, m; 1.57, m	1.82, m; 1.59, m
16	4.32, m	4.46, m	4.43, m	4.50, m	4.45, m
17	1.02, m	1.10, m	1.05, m	1.79, dt (7.3, 8.4)	1.72, m
18	0.88, s	0.83, s	0.83, s	0.81, s	0.83, s
19	0.83, s	0.87, s	0.63, s	0.70, s	0.66, s
20	1.70, m	1.95, m	2.04, m	1.94, m	1.80, m
21	1.11, d (6.6)	1.15, d (6.9)	1.15, d (6.6)	1.14, d (7.0)	1.15, d (6.9)
22	-	-	-	-	-
23	1.37, m; 1.32, m	2.02, m; 1.49, m	1.36, m; 1.34, m	1.53, m	1.57, m
24	1.83, m; 1.78, m	1.84, m; 1.79, m	0.90, m; 1.19, m	1.49, m	1.52, m
25	-	1.57, m	1.01, m	0.94, m	1.02, m
26	4.44, m; 4.02, m	3.60, m; 3.51, m	4.24, m; 3.88, m	3.61, m; 3.51, m	3.62, m; 3.52, m
27	4.84, s; 4.80, s	0.69, s	0.70, s	0.69, s	0.70, s
Ac-CH ₃	-	-	-	-	2.05, s

Table 1 ¹H NMR data (600 MHz) for the aglycone moieties of compounds 1-5^a (Pyridine-d₅, J in Hz)

^a Assignments are based on APT, HMQC, COSY, HMBC, and NOESY experiments.

HR-ESI-MS {*m*/*z* 1068.4167 [M + NH₄]⁺ (Calcd. 1068.4170)}. Analysis of its ¹H NMR spectrum (Table 1) displayed the signals of four anomeric protons at $\delta_{H}4.95$ (1H, d, J = 7.8 Hz), 5.62 (1H, d, J = 7.9 Hz), 5.21 (1H, d, J = 7.9 Hz), 5.34 (1H, d, J = 7.9 Hz) of four sugar moieties, and four methyl singlet signals at $\delta_{H}0.81$ (3H, s, Me-18), 0.70 (3H, s, Me-19), 1.14 (3H, d, J = 7.0 Hz, Me-21) and 0.69 (3H, s, Me-27) of aglycone. The ¹³C NMR spectrum (Table 2) showed a total of 50 carbon signals, where 27 carbons were assigned to the aglycone and the remaining 23 carbons to the sugar moieties. Actually, the ¹H and ¹³C NMR spectroscopic data of aglycone in **4** were similar to those of compound **3**, except for the hydroxyl position in the structure. The position of the hy-

droxyl group at C-2 in compound **4** was confirmed by the HMBC correlations of C-2 to H-1 ($\delta_{\rm H}$ 1.19, 2.20) and H-4 ($\delta_{\rm H}$ 1.66, 1.39), and the hydroxyl group was α -oriented based on the NOESY cross-peak of H-2 with H-19 ^[5]. The 25*R* form was determined by the chemical shifts of C-24 ($\delta_{\rm C}$ 28.99), C-25 ($\delta_{\rm C}$ 31.51) and C-26 ($\delta_{\rm C}$ 67.76) ^[3]. The 23 carbon signals were supposed to be four sugars, namely one D-xylose and three D-glucose moieties. The connectivity of the sugars was mainly based on the HMBC correlations: H-1' ($\delta_{\rm H}$ 4.95, d, *J* = 7.8 Hz) with C-3 ($\delta_{\rm C}$ 82.28) of the aglycone; H-1" ($\delta_{\rm H}$ 5.62, d, *J* = 7.9 Hz) with C-3' ($\delta_{\rm C}$ 84.97); H-1"' ($\delta_{\rm H}$ 5.21, d, *J* = 7.9 Hz) with C-3'' ($\delta_{\rm C}$ 80.69). All the sugars were determined to be β -oriented



No.	1	2	3	4	5	No.	1	2	3	No.	4	5
1	37.93	37.92	36.69	46.50	41.67	Gal-1'	103.63	103.62	103.29	Glc-1'	104.13	103.56
2	29.89	30.89	22.16	71.34	71.54	2'	76.46	76.45	76.50	2'	76.25	76.27
3	81.19	81.18	81.22	82.28	82.00	3'	74.13	74.13	74.14	3'	84.97	82.05
4	40.12	40.11	35.69	33.02	33.03	4'	82.34	82.01	82.05	4'	71.70	71.76
5	141.88	141.86	41.02	45.43	45.51	5'	78.49	78.48	78.48	5'	79.39	78.59
6	122.54	122.55	29.81	30.17	30.17	6'	63.97	63.97	63.97	6'	63.47	64.02
7	33.05	32.70	32.70	33.09	33.29	Glc-1"	106.07	106.07	106.06	Glc-1"	105.82	106.27
8	32.51	32.51	36.12	35.44	36.12	2"	79.59	79.59	79.59	2"	80.69	79.60
9	42.74	45.50	45.50	55.20	55.27	3"	89.40	89.37	89.39	3"	89.54	89.11
10	38.35	38.35	38.07	37.75	36.69	4"	71.75	71.75	71.75	4"	72.14	72.44
11	31.05	31.04	30.80	22.31	22.17	5"	79.64	79.63	79.63	5"	79.60	79.54
12	77.75	78.19	78.19	41.66	41.03	6"	61.41	61.41	61.50	6"	61.46	61.35
13	41.36	41.34	41.67	40.92	38.07	Glc-1'"	105.49	105.48	105.48	Xyl-1'''	105.48	105.48
14	51.11	55.23	55.24	57.19	57.33	2'''	76.18	76.17	75.84	2""	76.40	76.31
15	33.14	33.14	33.28	34.98	35.59	3'''	77.13	77.13	77.13	3'''	78.46	78.29
16	82.42	82.42	82.43	82.03	81.50	4'''	63.90	63.90	63.91	4'''	73.47	76.19
17	57.49	57.50	57.31	63.85	63.90	5'''	78.78	78.77	78.79	5'''	63.91	65.18
18	17.25	17.27	17.54	17.52	17.54	6'''	63.21	63.22	63.22	-	-	-
19	20.30	20.29	13.20	14.31	13.18	Glc-1""	105.87	105.86	105.86	Glc-1""	105.62	105.63
20	40.68	42.86	42.88	42.87	42.88	2""	76.23	76.22	76.22	2""	76.6	76.75
21	15.94	15.98	15.98	15.96	15.98	3""	72.47	72.48	72.48	3""	76.97	77.08
22	110.36	110.19	110.15	110.15	110.15	4""	63.76	63.75	63.72	4''''	72.48	74.27
23	21.98	33.09	33.03	32.69	32.70	5""	79.01	79.01	78.92	5''''	79.14	78.48
24	34.09	30.17	30.17	28.99	29.80	6""	63.18	63.17	63.16	6''''	63.20	63.17
25	145.33	31.51	31.51	31.51	31.51							
26	65.78	67.76	67.76	67.76	67.76							
27	109.69	18.26	18.26	18.26	18.26							
Ac-CO	-	-	-	-	172.14							
Ac-CH ₃	-	-	-	-	22.04							

Table 2 ¹³C NMR data (150 MHz) for compounds 1–5^a (Pyridine-d₅)

^a Assignments are based on APT, HMQC and HMBC experiments.

according to the *J* values. To sum up, compound **4** was identified as $25R-5\alpha$ -spirostane- 2α , 3β -diol- $3-O-\beta$ -D-glucopyranosyl(1 \rightarrow 2)-[β -D-xylopyranosyl (1 \rightarrow 3)]- β -D-glucopyranosyl (1 \rightarrow 3)- β -D-glucopyranoside.

Compound 5 was obtained as white powder. Its physicochemical properties were the same with that of compound 4, and its molecular formula was assigned as $C_{52}H_{84}O_{24}$ according to HR-ESI-MS. A comparison of the ¹H and ¹³C NMR spectroscopic data of 5 and 4 (Tables 1 and 2) suggested that they were similar except for the presence of acetoxyl group at δ_C 172.14 and 22.04 in 5. The position of the acetoxyl group at C-2 was confirmed by the HMBC correlations between C- Ac and H-2 ($\delta_{\rm H}4.01$), and the acetoxyl group was α -orientated based on the NOESY cross-peak of H-2 with H-19 ^[5]. Comparison of NMR spectrum for the sugar moieties of **5** with those of **4** suggested the same sugar chains, and the difference was mainly the substituents at C-2. Therefore, compound **5** was established as $25R-5\alpha$ -spirostane- 2α -O-Ac-3-O- β -D-glucopyranosyl(1 \rightarrow 2)-[β -D-xylopyranosyl(1 \rightarrow 3)]- β -Dglucopyranosyl(1 \rightarrow 3)- β -D-glucopyranoside.

Additionally, seven known compounds were isolated. By comparing their NMR spectroscopic data with literature values, their structures were determined to be coniferin (6) ^[18], naphthisoxazol A (7) ^[19], stagonoculiazin (8) ^[20], adenosine



Fig. 2 Key ¹H-¹H COSY, HMBC, and NOESY correlations of compounds 1–5

(9) $^{[21]}$, syringin (10) $^{[22]}$, tyramine (11) and 3-indolebutyric acid (12).

The antioxidant activities of these isolated monomer compounds were evaluated *in vitro* against DPPH radical, ABTS radical and OH radical (Fig. 3). Compounds 6–11 showed certain scavenging capacity for ABTS radical in a dose-dependent manner. The IC_{50} values of compounds **8–10** for ABTS radical scavenging were 1.367, 1.169 and 1.423 mg·mL⁻¹, respectively. Compounds **6**, **7** and **11** had enhanced ABTS radical scavenging capacity, with IC_{50} values







of 0.208, 0.057 and 0.014 mg \cdot mL⁻¹, respectively.

Experimental

General experimental procedures

NMR spectra were obtained using a Bruker AV III 600 NMR spectrometer with chemical shift values presented as δ values, with TMS as the internal standard (samples dissolved in an appropriate amount of deuterated pyridine- d_5 and deuterated methanol). HR-ESI-MS was performed using an LTQ-Orbitrap XL spectrometer (Thermo Fisher Scientific, Boston, MA, USA). Acchrom S6000 high-performance liquid chromatograph [Acchrom Tech (Beijing) Technology Co., Ltd., Beijing, China] equipped with ELSD-UM 5800 Plus [Unimicro (Shanghai) Technologies Co., Ltd., Shanghai, China] was used for analysis. Colurmn chromatography (CC) was performed using silica gel (100-200 and 200-300 mesh, Qingdao Marine Chemical Plant, Qingdao, China), inverted C₁₈ filler (50 µm, Nantong Hairuo Chemical Co., Ltd., Nantong, China) and macroporous resin D101 (Tianjin Haoju Resin Technology Co., Ltd., Tianjin, China). All solvents used (petroleum ether, ethanol, methanol, n-butanol, dichloromethane, deuterated pyridine- d_5 and deuterated methanol) were of analytical grade (Beijing Chemical Plant, Beijing, China). Plant material

Fresh Allii Macrostemonis Bulbus was purchased from Changchun, and dried in an oven at 50 °C. The dried samples were crushed, sieved through 40 mesh sieve, sealed and stored for later use.

Extraction and isolation

The dried powder of Allii Macrostemonis Bulbus (4.0 kg) was extracted by petroleum ether under reflux (3 times, 2 h each time) at a solid-liquid ratio of 1 : 5 (W/V) to degrease. According to the solid-liquid ratio of 1 : 10 (W/V), the filter residue was extracted with 75% ethanol solution (V/V, 1 h each time) under reflux for three times. After removal of the solvent, extraction was performed using saturated *n*-butanol for three times, and the resultant extract was subjected to D-101 macroporous resin column using a gradient of EtOH–H₂O to give four fractions (30%, 50%, 70% and 90%).

The 90% fraction of the above-mentioned (12.795 g) was subjected to a silica gel (100–200 mesh) column, eluting with a stepwise gradient of CH₂Cl₂/MeOH (from 9 : 1 to 0 : 1, V/V) to give seven fractions (Frs. A–G) by TLC analysis. Fr.C (2.32 g) was loaded onto a silica gel column (200–300 mesh) and eluted with CH₂Cl₂/MeOH/H₂O (stepwise, from 9 : 3 : 1 to 7 : 4 : 1, V/V/V) to yield five fractions (Frs. C-1–5). Fr. C-3 (0.87 g) was separated over an ODS (50 µm) column, eluting with MeOH/H₂O (stepwise, from 3 : 2 to 9 : 1, V/V) to give compound **2** (134.81 mg) and compound **3** (145.64 mg). Fr. C-5 (0.51 g) was subjected to an ODS (50 µm) column and eluted with MeOH/H₂O (stepwise, from 3 : 2 to 9 : 1, V/V) to yield compound **1** (86.68 mg). Fr. E (1.12 g) was applied onto a silica gel (200–300 mesh) column, eluting with CH₂Cl₂/MeOH/H₂O (stepwise, from

8:3:1 to 5:3:1, V/V/V to obtain four fractions (Frs. E-1-4). Fr. E-2 was further subjected to an ODS (50 µm) column and eluted with MeOH/H2O (stepwise, from 4 : 1 to 9:1, V/V to give compound 5 (11.18 mg). Fr. E-3 was subjected to an ODS (50 µm) column and eluted with MeOH/H₂O (stepwise, from 4 : 1 to 9 : 1, V/V) to vield compound 4 (9.85 mg). The 30% fraction (16.007 g) was subjected to a silica gel (100-200 mesh) column, eluting with a stepwise gradient of $CH_2Cl_2/MeOH$ (from 7 : 1 to 0 : 1, V/V) to give nine fractions (Frs. H-P) by TLC analysis. Fr. I (0.42 g) was separated over a silica gel (200-300 mesh) column, eluting with CH₂Cl₂/MeOH/H₂O (stepwise, from 11:4:2 to 6.5 : 2.5 : 1, V/V/V) to yield five fractions (Frs. I-1–5). Fr. I-1 was loaded onto an ODS (50 µm) column and eluted with ACN/H₂O (1 : 9, V/V) to obtain compound 6 (10.74 mg) Fr. I-3 was subjected to an ODS (50 µm) column and eluted with ACN/H₂O (1 : 9, V/V) to yield compound 10 (22.90 mg) and compound 8 (10.14 mg). Fr. K was recrystallized to give compound 9 (123.55 mg). Fr. O (1.13 g) was chromatographed over a silica gel column (200-300 mesh), eluting with CH₂Cl₂/MeOH/H₂O (stepwise, from 10:4:2 to 13:8:1, V/V/V to obtain eight fractions (Frs. O-1-8). Fr. O-5 was applied onto an ODS (50 µm) column and eluted with MeOH/H₂O (stepwise, from $1 \div 5$ to $3 \div 7$, V/V) to yield compound 7 (73.75 mg). Fr. O-6 was subjected to an ODS (50 µm) column, eluting with MeOH/H2O (stepwise, from 1:5 to 3:7, V/V) to give compound 11 (13.33 mg) and compound 12 (7.07 mg).

Antioxidant assay

2,2-Diphenyl-1-picrylhydrazylradical (DPPH) scavenging assav

The DPPH radical scavenging capacity was determined as described by Om P *et al.* ^[23]. Compounds **1–11** and Vc (50 μ L, 0–2 mg·mL⁻¹) were separately mixed with fresh DPPH ethanol solution (150 μ L, 0.4 mmol·L⁻¹), and then allowed to react in the darkness for 30 min. V_C was used as a positive control. The absorbance was measured at 517 nm. The DPPH radical scavenging rate was calculated according to formula (1):

Scavenging rate (%) = $[1 - (A_1 - A_2)/A_0] \times 100$ (1)

Where A_0 was the absorbance of DPPH solution (150 μ L) with distilled water (50 μ L); A_1 was the absorbance of compounds **1–11**/Vc (50 μ L, 0–2 mg·mL⁻¹) and DPPH solution (150 μ L); and A_2 was the absorbance of absolute ethanol instead of DPPH solution.

2,2'-Azino-bis (3-ethylbenzothiazoline-6-sulfonic acid) ABTS) scavenging assay

Herein, 204 mg of ABTS and 35.14 mg of $K_2S_2O_8$ were separately diluted to 50 mL with water, mixed and shaked well, avoiding light reaction at room temperature for 16 h to yield ABTS mother liquor. The above solution was diluted with ethanol until its absorbance was 0.7 at 734 nm. Samples and Vc (80 µL, 0–2 mg·mL⁻¹) were separately mixed with ABTS solution (140 µL), and then allowed to react in the darkness at room temperature for 30 min. The absorbance was measured at 734 nm. The ABTS radical scavenging rate was calculated according to formula (2):

Scavenging rate (%) = $(1 - A_1/A_0) \times 100$ (2)

Where A_0 was the absorbance of ABTS solution (140 μ L) with distilled water (80 μ L); A_1 was the absorbance of compounds 1–11/Vc (80 μ L, 0–2 mg·mL⁻¹) and ABTS solution (140 μ L)^[24].

Hydroxyl radical (•OH) scavenging assay

Salicylic acid ethanol solution (50 μ L, 10 mmol·L⁻¹), FeSO₄ solution (50 μ L, 10 mmol·L⁻¹) and compound 1–11/Vc (50 μ L, 0–2 mg·mL⁻¹) were absorbed to be tested respectively in 96-well plate, mixed well and reacted at 37 °C for 30 min. H₂O₂ solution (50 μ L, 100 mmol·L⁻¹) was added to terminate the reaction. The absorbance was measured at 510 nm. The OH radical scavenging rate was calculated according to formula (3):

Scavenging rate (%) =
$$[1 - (A_1 - A_2)/A_0] \times 100$$
 (3)

Where A_0 was the absorbance of salicylic acid ethanol solution (50 µL), FeSO₄ solution (50 µL), distilled water (50 µL) and H₂O₂ solution (50 µL); A_1 was the absorbance of compounds **1–11**/Vc (50 µL, 0–2 mg·mL⁻¹), salicylic acid ethanol solution (50 µL), FeSO₄ solution (50 µL) and H₂O₂ solution (50 µL); and A_2 was the absorbance of distilled water instead of H₂O₂ solution ^[25].

Supporting Information

The NMR spectra of compounds 1–12 are available as Supporting Information, and can be requested by sending Email to the corresponding author.

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